# organic papers

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# Li-Ping Gao,<sup>a,b</sup>\* Ping-Dong Wu,<sup>a</sup> Xiu-Rong Hu<sup>c</sup> and Jian-Ming Gu<sup>c</sup>

<sup>a</sup>Institute of Pharmaceutical Engineering, College of Materials and Chemical Engineering, Zhejiang University, Hangzhou, Zhejiang 310027, People's Republic of China, <sup>b</sup>Department of Applied Chemistry, Zhejiang Science and Technical University, Hangzhou, Zhejiang 310017, People's Republic of China, and <sup>c</sup>Center of Analysis and Measurement, Zhejiang University, Hangzhou, Zhejiang 310028, People's Republic of China

Correspondence e-mail: gaoliping2002@hotmail.com

#### **Key indicators**

Single-crystal X-ray study T = 296 KMean  $\sigma$ (C–C) = 0.004 Å R factor = 0.036 wR factor = 0.060 Data-to-parameter ratio = 8.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# a Section E (8S)-3-(2-Deoxy orts orwthere points

# (8S)-3-(2-Deoxy-3,5-di-O-p-toluoyl-a-Derythro-pentofuranosyl)-3,6,7,8-tetrahydroimidazo[4,5-d][1,3]diazepin-8-ol

In the asymmetric unit of the title compound,  $C_{27}H_{28}N_4O_6$ , there are two independent molecules which have different conformations. The tetrahydrofuran and dihydrodiazepine rings adopt envelope conformations. The hydroxy, amino and carbonyl groups are all involved in hydrogen bonding.

### Comment

The title compound, (I), is an intermediate in the synthesis of pentostatin, which is an antimetabolite/antineoplastic agent, one of the newest chemotherapy drugs (Kasibhatla & Erion, 2000; Kasibhatla *et al.* 2001).



Compound (I) crystallizes with two independent molecules, A and B, in the asymmetric unit. These two molecules have different conformations (Fig. 1 and Table 1), and have different hydrogen-bonding characteristics (see below). In molecule A, the carbonyl group C12=O14 deviates from the least-squares plane of the C13-C18 aromatic ring, as shown by the O14-C12-C13-C14 torsion angle of 23.2 (5) Å. Iy contrast, in molecule B, the O34-C42-C43-C44 torsion angle is  $2.5 (5)^{\circ}$ , indicating coplanarity. In addition, there are significant differences in the relative orientations of the C<sub>6</sub>aromatic rings of the tolyl groups in molecules A and B, as seen by the dihedral angles between them of 30.33 (13) and 22.85 (13)°, respectively. Furthermore, in molecule A, the imidazole ring plane is almost parallel to the C21-C26 aromatic ring, forming a dihedral angle of  $5.75 (14)^{\circ}$ , but forming a dihedral angle of 29.76 (13)° with the second

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#### Figure 1

The molecular structures of the two indpendent molecules comprising the asymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.



#### Figure 2

A packing diagram for (I). Dashed lines indicate hydrogen bonds. [Symmetry codes: (i)  $1 - x, \frac{1}{2} + y, 1 - z$ ; (ii)  $1 - x, \frac{1}{2} + y, 1 - z$ .]

The hydroxy, amino and carbonyl groups are all involved in the hydrogen-bonding scheme, as detailed in Table 2. N- $H \cdots N$  hydrogen bonds link the molecules into a linear hydrogen-bonded ribbon motif along the *b*-axis direction and this chain is further stabilized by  $O-H\cdots O$  interactions (Fig. 2).

# **Experimental**

To 3-(2-deoxy-3,5-di-O-p-toluoyl-α-D-erythro- pentafuranosyl)-6,7dihydroimidazo[4,5-*d*][1,3]diazepin-8(3*H*)-one (5.0 g, 0.01 mol) dissolved in MeOH (150 ml) was added sodium borohydride (0.15 g, 0.004 mol). The solution was stirred at 298 K for 0.5 h, at the end of which time the excess reducing agent was decomposed by the addition of dry ice. High-performance liquid chromatography revealed a ca 1:1 mixture of (8S)-3-(2-deoxy-3,5-di-O-p-toluoyl-α-D-erythropentafuranosyl)-3,6,7,8-tetrahydroimidazo[4,5-d][1,3]diazepin-8-ol and (8R)-3-(2-deoxy-3,5-di-O-p-toluoyl-a-D-erythro-pentafuranosyl-3,6,7,8-tetrahydroimidazo[4,5-*d*][1,3]diazepin-8-ol. The above diastereoisomeric mixture was decolorized using active carbon and filtered. Evaporation of the filtrate left 4.2 g of a light-brown solid residue. This solid was dissolved in hot ethyl acetate (150 ml) and the solution was left to stand at 273 K for 3 d. The crystalline 8S-isomer was filtered off and washed with cold ethyl acetate. Recrystallization three times gave 99% analytically pure 8S-isomer (Baker & Putt, 1979; Showalter & Putt, 1981; Chan et al., 1982). Compound (I) was recrystallized from an acetone solution yielding yellow crystals suitable for the X-ray diffraction study.

#### Crystal data

$C_{27}H_{28}N_4O_6$	Z = 4
$M_r = 504.54$	$D_x = 1.346 \text{ Mg m}^{-3}$
Monoclinic, P2 <sub>1</sub>	Mo $K\alpha$ radiation
a = 10.492 (2)  Å	$\mu = 0.10 \text{ mm}^{-1}$
b = 12.849 (3)  Å	T = 296 (1)  K
c = 18.645 (5)  Å	Block, yellow
$\beta = 97.949 \ (10)^{\circ}$	$0.30 \times 0.14 \times 0.12 \text{ mm}$
$V = 2489.5 (10) \text{ Å}^3$	

# Data collection

Rigaku R-AXIS RAPID	5949 independent reflections
diffractometer	2974 reflections with $F^2 > 2\sigma$
$\omega$ scans	$R_{\rm int} = 0.044$
Absorption correction: none	$\theta_{\rm max} = 27.5^{\circ}$
24329 measured reflections	

#### Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma(F^2)] = 0.036$ wR(F<sup>2</sup>) = 0.060 S = 1.005949 reflections 668 parameters H-atom parameters constrained  $F^2$ )

 $w = 1/[0.8800\sigma(F_0^2)]/(4F_0^2)$  $(\Delta/\sigma)_{\rm max} = 0.001$  $\Delta \rho_{\rm max} = 0.32 \ {\rm e} \ {\rm \AA}$  $\Delta \rho_{\rm min} = -0.34 \text{ e } \text{\AA}^{-3}$ Extinction correction: Larson (1970), equation 22 Extinction coefficient: 198 (9)

Table 1	
Selected geometric parameters (Å,	°).

O13-C11	1.437 (3)	N13-C9	1.344 (4)
O13-C12	1.354 (3)	N14-C9	1.274 (3)
O14-C12	1.193 (4)	N14-C10	1.382 (3)
O15-C20	1.341 (3)	N32-C36	1.378 (3)
O33-C41	1.455 (3)	N33-C39	1.339 (4)
O33-C42	1.335 (3)	N34-C39	1.292 (3)
O34-C42	1.208 (4)	N34-C40	1.368 (3)
O35-C50	1.356 (3)	C6-C7	1.486 (3)
N12-C6	1.392 (3)	C36-C37	1.505 (4)
C5-N11-C4-O11	-68.0(3)	O13-C12-C13-C18	24.0 (4)
C10-N11-C4-O11	116.9 (3)	O14-C12-C13-C14	23.2 (5)
C35-N31-C34-O31	-85.1(3)	O33-C42-C43-C48	3.0 (4)
C40-N31-C34-O31	91.9 (3)	O34-C42-C43-C44	2.5 (5)

 Table 2

 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O12-H121O34 <sup>i</sup>	0.91	1.98	2.889 (3)	173
O32-H321···O16	0.91	2.23	3.013 (3)	144
$N13-H130\cdots N12^{i}$	0.86	2.22	2.975 (3)	146
$N33-H330\cdots N32^{ii}$	0.86	2.33	2.954 (3)	130

Symmetry codes: (i) -x + 1,  $y - \frac{1}{2}$ , -z + 1; (ii) -x + 2,  $y + \frac{1}{2}$ , -z + 2.

O-bound H atoms were located from difference Fourier maps and included in the refinement based on the as-found O–H bond lengths (O–H = 0.91 Å), and with  $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}$ (parent atom). The N– and C-bound H atoms were included in the riding-model approximation, with N–H = 0.86 Å and C–H = 0.93–0.98 Å, and  $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}$ (parent atom). In the absence of significant anomalous scat-

tering effects, 4005 Friedel pairs were averaged in the final refinement.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/ MSC, 2004); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure*.

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