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## Key indicators

Single-crystal X-ray study
$T=296 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.036$
$w R$ factor $=0.060$
Data-to-parameter ratio $=8.9$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
# (8S)-3-(2-Deoxy-3,5-di-O-p-toluoyl- $\alpha$-D-erythro-pentofuranosyl)-3,6,7,8-tetrahydro-imidazo[4,5-d][1,3]diazepin-8-ol 

In the asymmetric unit of the title compound, $\mathrm{C}_{27} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{6}$, there are two independent molecules which have different conformations. The tetrahydrofuran and dihydrodiazepine rings adopt envelope conformations. The hydroxy, amino and carbonyl groups are all involved in hydrogen bonding.

## Comment

The title compound, (I), is an intermediate in the synthesis of pentostatin, which is an antimetabolite/antineoplastic agent, one of the newest chemotherapy drugs (Kasibhatla \& Erion, 2000; Kasibhatla et al. 2001).

(I)

Compound (I) crystallizes with two independent molecules, $A$ and $B$, in the asymmetric unit. These two molecules have different conformations (Fig. 1 and Table 1), and have different hydrogen-bonding characteristics (see below). In molecule $A$, the carbonyl group $\mathrm{C} 12=\mathrm{O} 14$ deviates from the least-squares plane of the C13-C18 aromatic ring, as shown by the $\mathrm{O} 14-\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 14$ torsion angle of 23.2 (5) $\AA$. Iy contrast, in molecule $B$, the $\mathrm{O} 34-\mathrm{C} 42-\mathrm{C} 43-\mathrm{C} 44$ torsion angle is $2.5(5)^{\circ}$, indicating coplanarity. In addition, there are significant differences in the relative orientations of the $\mathrm{C}_{6}$ aromatic rings of the tolyl groups in molecules $A$ and $B$, as seen by the dihedral angles between them of 30.33 (13) and $22.85(13)^{\circ}$, respectively. Furthermore, in molecule $A$, the imidazole ring plane is almost parallel to the C21-C26 aromatic ring, forming a dihedral angle of $5.75(14)^{\circ}$, but forming a dihedral angle of $29.76(13)^{\circ}$ with the second

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Figure 1
The molecular structures of the two indpendent molecules comprising the asymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level.


Figure 2
A packing diagram for (I). Dashed lines indicate hydrogen bonds. [Symmetry codes: (i) $1-x,-\frac{1}{2}+y, 1-z$; (ii) $1-x, \frac{1}{2}+y, 1-z$.]
aromatic ring containing $\mathrm{C} 13-\mathrm{C} 18$. In contrast, in molecule $B$, the imidazole ring is almost parallel to the C43-C48 aromatic ring [dihedral angle $7.57(12)^{\circ}$ ] but forms a dihedral angle of 26.42 (13) ${ }^{\circ}$ with the C51-C56 ring, i.e. opposite to what is observed for molecule $A$. In both independent molecules, the tetrahydrofuran and dihydrodiazepine rings adopt envelope conformations.

The hydroxy, amino and carbonyl groups are all involved in the hydrogen-bonding scheme, as detailed in Table 2. N$\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds link the molecules into a linear hydrogen-bonded ribbon motif along the $b$-axis direction and this chain is further stabilized by $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ interactions (Fig. 2).

## Experimental

To 3-(2-deoxy-3,5-di- $O$ - $p$-toluoyl- $\alpha$-D-erythro- pentafuranosyl)-6,7dihydroimidazo $[4,5-d][1,3]$ diazepin- $8(3 H)$-one $\quad(5.0 \mathrm{~g}, \quad 0.01 \mathrm{~mol})$ dissolved in $\mathrm{MeOH}(150 \mathrm{ml})$ was added sodium borohydride $(0.15 \mathrm{~g}$, 0.004 mol ). The solution was stirred at 298 K for 0.5 h , at the end of which time the excess reducing agent was decomposed by the addition of dry ice. High-performance liquid chromatography revealed a ca 1:1 mixture of (8S)-3-(2-deoxy-3,5-di-O-p-toluoyl- $\alpha$-D-erythro-pentafuranosyl)-3,6,7,8-tetrahydroimidazo $[4,5-d][1,3]$ diazepin-8-ol and ( $8 R$ )-3-(2-deoxy-3,5-di-O-p-toluoyl- $\alpha$-D-erythro-pentafuranosyl-$3,6,7,8$-tetrahydroimidazo $[4,5-d][1,3]$ diazepin- 8 -ol. The above diastereoisomeric mixture was decolorized using active carbon and filtered. Evaporation of the filtrate left 4.2 g of a light-brown solid residue. This solid was dissolved in hot ethyl acetate $(150 \mathrm{ml})$ and the solution was left to stand at 273 K for 3 d . The crystalline $8 S$-isomer was filtered off and washed with cold ethyl acetate. Recrystallization three times gave $99 \%$ analytically pure $8 S$-isomer (Baker \& Putt, 1979; Showalter \& Putt, 1981; Chan et al., 1982). Compound (I) was recrystallized from an acetone solution yielding yellow crystals suitable for the X-ray diffraction study.

## Crystal data

$\mathrm{C}_{27} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{6}$
$M_{r}=504.54$
Monoclinic, $P 2_{1}$
$a=10.492$ (2) A
$b=12.849$ (3) $\AA$
$c=18.645$ (5) $\AA$
$\beta=97.949$ (10) ${ }^{\circ}$
$V=2489.5(10) \AA^{3}$

## Data collection

Rigaku R-AXIS RAPID
diffractometer
$\omega$ scans
Absorption correction: none
24329 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.036$
$w R\left(F^{2}\right)=0.060$
$S=1.00$
5949 reflections
668 parameters
H -atom parameters constrained
$Z=4$
$D_{x}=1.346 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=0.10 \mathrm{~mm}^{-1}$
$T=296$ (1) K
Block, yellow
$0.30 \times 0.14 \times 0.12 \mathrm{~mm}$

> 5949 independent reflections 2974 reflections with $F^{2}>2 \sigma\left(F^{2}\right)$
> $R_{\text {int }}=0.044$
> $\theta_{\max }=27.5^{\circ}$

[^1]Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| O13-C11 | $1.437(3)$ | N13-C9 | $1.344(4)$ |
| :--- | ---: | :--- | ---: |
| O13-C12 | $1.354(3)$ | N14-C9 | $1.274(3)$ |
| O14-C12 | $1.193(4)$ | N14-C10 | $1.382(3)$ |
| O15-C20 | $1.341(3)$ | N32-C36 | $1.378(3)$ |
| O33-C41 | $1.455(3)$ | N33-C39 | $1.339(4)$ |
| O33-C42 | $1.335(3)$ | N34-C39 | $1.292(3)$ |
| O34-C42 | $1.208(4)$ | N34-C40 | $1.368(3)$ |
| O35-C50 | $1.356(3)$ | C6-C7 | $1.486(3)$ |
| N12-C6 | $1.392(3)$ | C36-C37 | $1.505(4)$ |
|  |  |  |  |
| C5-N11-C4-O11 | $-68.0(3)$ | O13-C12-C13-C18 | $24.0(4)$ |
| C10-N11-C4-O11 | $116.9(3)$ | O14-C12-C13-C14 | $23.2(5)$ |
| C35-N31-C34-O31 | $-85.1(3)$ | O33-C42-C43-C48 | $3.0(4)$ |
| C40-N31-C34-O31 | $91.9(3)$ | O34-C42-C43-C44 | $2.5(5)$ |

Table 2
Hydrogen-bond geometry ( $\AA,{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| O12-H121 $\cdots \mathrm{O} 34^{\mathrm{i}}$ | 0.91 | 1.98 | $2.889(3)$ | 173 |
| O32-H321 $\cdots$ O16 | 0.91 | 2.23 | $3.013(3)$ | 144 |
| N13-H130 $\mathrm{N}^{\mathrm{i}}$ | 0.86 | 2.22 | $2.975(3)$ | 146 |
| N33-H330 $\cdots \mathrm{N} 32^{\mathrm{ii}}$ | 0.86 | 2.33 | $2.954(3)$ | 130 |

Symmetry codes: (i) $-x+1, y-\frac{1}{2},-z+1$; (ii) $-x+2, y+\frac{1}{2},-z+2$.
O-bound H atoms were located from difference Fourier maps and included in the refinement based on the as-found $\mathrm{O}-\mathrm{H}$ bond lengths $(\mathrm{O}-\mathrm{H}=0.91 \AA)$, and with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}$ (parent atom). The $\mathrm{N}-$ and C -bound H atoms were included in the riding-model approximation, with $\mathrm{N}-\mathrm{H}=0.86 \AA$ and $\mathrm{C}-\mathrm{H}=0.93-0.98 \AA$, and $U_{\text {iso }}(\mathrm{H})=$ $1.2 U_{\text {eq }}$ (parent atom). In the absence of significant anomalous scat-
tering effects, 4005 Friedel pairs were averaged in the final refinement.

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/ MSC, 2004); program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: CRYSTALS (Betteridge et al., 2003); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: CrystalStructure.

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## References

Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. \& Spagna, R. (1999). J. Appl. Cryst. 32, 115-119.
Baker, D. C. \& Putt, S. R. (1979). J. Am. Chem. Soc. 101, 6127-6128.
Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. \& Watkin, D. J. (2003). J. Appl. Cryst. 36, 1487.

Chan, E., Putt, S. R. \& Showalter, H. D. H. (1982). J. Org. Chem. 47, 34573464.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Kasibhatla, S. R., Bookser, B. C., Xiao, W. \& Erion, M. D. (2001). J. Med. Chem. 44, 613-618.
Kasibhatla, S. R. \& Erion, M. D. (2000). J. Med. Chem. 43, 1519-1524.
Larson, A. C. (1970). Crystallographic Computing, edited by F. R. Ahmed, S. R. Hall \& C. P. Huber, pp. 291-294. Copenhagen: Munksgaard.

Rigaku (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.
Rigaku/MSC (2004). CrystalStructure. Version 3.6.0. Rigaku/MSC, The Woodlands, Texas, USA.
Showalter, H. D. H. \& Putt, S. R. (1981). Tetrahedron Lett. Vol.22, 3155-3158.


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[^1]:    $w=1 /\left[0.8800 \sigma\left(F_{\mathrm{o}}{ }^{2}\right)\right] /\left(4 F_{\mathrm{o}}{ }^{2}\right)$
    $(\Delta / \sigma)_{\text {max }}=0.001$
    $\Delta \rho_{\text {max }}=0.32 \mathrm{e}^{-3} \mathrm{~A}^{-3}$
    $\Delta \rho_{\text {min }}=-0.34 \mathrm{e}^{-3}$
    Extinction correction: Larson
    (1970), equation 22

    Extinction coefficient: 198 (9)

